Note

New acyclic-sugar *C*-nucleoside analogs. Synthesis of 2-(alditol-1-yl)-1-alkyl-4,5,6,7-tetrahydroindol-4-ones

JUAN A. GALBIS PEREZ*, JOSE L. JIMENEZ REQUEJO, JUAN C. PALACIOS ALBARRAN, AND MARTIN AVALOS GONZALEZ

Department of Organic Chemistry, Faculty of Sciences, University of Extremadura, Badajoz (Spain) (Received June 2nd, 1984; accepted for publication, October 23rd, 1984)

The reaction between 2-amino-2-deoxyaldoses and cyclic 1,3-dicarbonyl compounds yields tetrahydro-(polyhydroxyalkyl)indol-4-ones^{1,2}. These acyclic-sugar *C*-nucleoside analogs can be converted into cyclic ones by acid-catalyzed dehydration of the polyhydroxyalkyl chain³⁻⁵. We now report on the preparation of new 2-(al-ditol-1-yl)-1-alkyl-4,5,6,7-tetrahydroindol-4-ones by reaction of 5,5-dimethyl-1,3-cyclohexanedione and 1,3-cyclohexanedione with 2-(alkylamino)-2-deoxyaldoses that we have previously described⁶⁻⁸.

Thus, by reaction of 2-deoxy-2-(ethylamino)-L-glucose⁶ with 5,5-dimethyl-1,3-cyclohexanedione, 1-ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(L-arabino-tetritol-1-yl)indol-4-one (1) was obtained in good yield. The structure of 1 was demonstrated by elemental analysis and spectral data (u.v. and i.r.). Periodate oxidation of the polyhydroxyalkyl chain gave a reagent consumption of 3 mol per mol of compound, indicative of four contiguous hydroxyl groups. This oxidation yields 1-ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-4-oxoindol-2-carboxaldehyde (12), which confirms the assigned structure of the heterocyclic ring. The presence of the tetrahydroxybutyl chain was also proved by preparing the tetra-O-acetyl derivative 2. Assignment of the L-arabino configuration is based on the configuration of the sugar precursor of 1, and is consistent with the positive value of its rotatory power, in agreement with generalized rules⁹⁻¹¹.

In the same way, the reaction of 2-deoxy-2-(ethylamino)-D-glycero-L-gluco-heptose⁸ with 5,5-dimethyl-1,3-cyclohexanedione and 1,3-cyclohexanedione afforded 1-ethyl-tetrahydro-2-(D-galacto-pentitol-1-yl)indol-4-ones 3 and 4, and the reaction of 2-deoxy-2-(propylamino)-D-glycero-L-gluco- or D-glycero-L-manno-heptose⁸ with 5,5-dimethyl-1,3-cyclohexanedione gave tetrahydro-2-(D-galacto-

^{*}Present address: Department of Organic Chemistry, Faculty of Pharmacy, University of Seville, Seville, Spain.

pentitol-1-yl)-1-propylindol-4-one (5). The structures assigned to these compounds were in agreement with their elemental analyses, spectral data (u.v. and i.r.). positive rotatory powers, and periodate oxidation of the pentahydroxypentyl chain, \sim 4 mol. equiv. of periodate being consumed in each case. Acetylation of these compounds gave the penta-O-acetyl derivatives 6-8. The structure of the heterocyclic ring system was also confirmed by oxidative degradation of the side chain of compounds 3-5, to give the corresponding carboxaldehydes (12-14).

The reaction of 2-deoxy-2-(ethylamino)-D-glycero-D-talo-heptose⁷ with 5,5-dimethyl-1,3-cyclohexanedione gave 1-ethyl-tetrahydro-2-(D-manno-pentitol-1-yl)indol-4-one (9), and reaction of the epimeric mixture of 2-deoxy-2-(propyl-amino)-D-glycero-D-ido- and D-glycero-D-gulo-heptose with the same 1,3-dicarbonyl compound led to tetrahydro-2-(D-gluco-pentitol-1-yl)-1-propylindol-4-one (11). The structures of compounds 9 and 11 were demonstrated by the methods described for compounds 3-5.

EXPERIMENTAL

General methods. — Solutions were evaporated in vacuo at temperatures below 50°. Melting points were determined with a Gallenkamp apparatus, and are uncorrected. Optical rotations were measured at 22 ±4° with a Perkin-Elmer 141 polarimeter (10-cm, 5-mL cell). I.r. spectra (KBr discs) were recorded with a Perkin–Elmer 399 spectrophotometer, and u.v. spectra with a Pye–Unicam SP8-250 instrument. Paper chromatography was performed on Whatman No. 1 paper by the ascending technique, with 1:1:1 1-butanol-pyridine-water as the eluant, and silver nitrate-sodium hydroxide or Ehrlich reagent as the indicator. T.l.c. was performed on silica gel GF₂₅₄ (Merck) with 3:1 ethyl acetate-ethanol as the eluant, and detection with u.v. light, iodine vapor, or Ehrlich reagent for pyrroles. Consumption of periodate was determined as previously described¹². ¹H-N.m.r. spectra at 90 MHz, with internal Me₄Si, were recorded with a Perkin-Elmer R-32 spectrometer, and coupling constants were measured directly from the spectra recorded at 300-Hz sweep-width (temperature of the probe, 35.5°). Assignments were confirmed by double resonance (spin-spin decoupling and spin-tickling), and overlapping signals were gradually shifted, and separated from one another, by incremental additions of Eu(fod)3.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(L-arabino-tetritol-1-yl)indol-4-one (1). — A solution of 2-deoxy-2-ethylamino-L-glucose hydrochloride⁶ (4.0 g, 15.3 mmol) in water (10 mL) was treated with 5,5-dimethyl-1,3-cyclohexanedione (2.3 g, 16.4 mmol) in 2:1 acetone-water (23 mL), and sodium carbonate (0.87 g, 82 mmol) was added. The mixture was kept for 14 days at room temperature, and then the acetone was evaporated under diminished pressure. The resulting solution was washed with chloroform (2 × 15 mL), and evaporated, to yield 1, which was filtered off, and successively washed with 1:1 ethanol-acetone, acetone, and ether (yield 3.8 g, 80%); recrystallization from acetone gave needles; m.p. 125-127°, $[\alpha]_D + 12^\circ$, $[\alpha]_{578} + 12.5^\circ$, $[\alpha]_{546} + 15^\circ$, $[\alpha]_{436} + 32^\circ$, $[\alpha]_{365} + 75^\circ$ (c 0.5, pyridine); $\lambda_{\text{max}}^{\text{EtOH}}$ 250 and 280 nm (ε_{mM} 8.80 and 6.30); ν_{max} 3600-3000 (OH), 1635 and 1615 (C=O), 1545, and 1505 cm⁻¹ (C=C pyrrole).

Anal. Calc. for $C_{16}H_{25}NO_5$: C, 61.72; H, 8.09; N, 4.50. Found: C, 61.67; H, 8.06; N, 4.64. Periodate consumption: 3.02 mol.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(tetra-O-acetyl-L-arabino-tetritol-1-yl)indol-4-one (2). — A solution of 1 (0.8 g, 2.5 mmol) in pyridine (3 mL) and acetic anhydride (3.5 mL) was kept for 24 h at room temperature, and then poured into ice-water (50 mL). The solid 2 (1.1 g, 86%) was filtered off, washed with water, and dried over sodium hydroxide; m.p. 53–55°, $[\alpha]_D$ +47°, $[\alpha]_{578}$ +49°, $[\alpha]_{546}$ +55.5°, $[\alpha]_{436}$ +107°, $[\alpha]_{365}$ +201° (c 1.0, chloroform); ν_{max} 1725 (C=O ester), 1630 (C=O ketone), 1540, and 1500 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table I.

Anal. Calc. for $C_{24}H_{33}NO_9 \cdot H_2O$: C, 57.94; H, 7.09; N, 2.82. Found: C, 57.90; H, 7.17; N, 2.96.

TABLE I

¹H-N M R DATA⁴ (90 MHz) FOR 2, 6, 7, 8, AND 10

Compound H-1'b H-2'	H-I'b	H-2′	Н-3′	H-4'	H-4" H-5'	Н-5′	Н-5" Не	Hr	CH_3^c	N-Et	N-Pr	-CH2' OAc	ОАс
7	$6.12\mathrm{d}^d$ $J_{1,2},4.7$	6.12 d ^d 5.44 dd $J_{1,2}$ 4.7 $J_{2,3}$ 6.7	5.30 m J _{3'.4'} 3.3	4.27 dd J _{4',4"} -12.7	4.08 dd			6.57s 111s	1 11 s	3.99 q (2 H) 1.37 t (3 H)		2.61 s 2.32 s	2.07 s (6 H) 2.03 s (6 H)
9	5.98 d J _{1',2'} 3.2	5.98 d 5.30 dd J _{1,2} , 3,2 J _{2,3} , 9.3	J _{3',4"} 5.5 5.52 dd J _{3',4'} 2.3	J _{3',4''} 5.5 5.52 dd 5.32 m J _{3',4'} 2.3 J _{4',5'} 5.0		4.28 dd J _{5.5"} -11.3	3.93 dd 6 49 s 1.10 s	6 49 s	1.10s	3.95 q (2 H) 1.39 t (3 H)		2.60 s 2.32 s	2.10 s (6 H) 2.04 s (6 H)
7	5.92 d J _{1'.2'} 3.3		5.48 dd J _{3'.4''} 2.0	J _{4'.5'} 2.6 5.30 m J _{4'} 5.5.3		$4.24 dd$ $J_{5'.5''} - 12.0$	3.85 dd 6.47 s	6.47 s		3.91 q (2 H) 1.36 t (3 H)		2.70 m 2.35 m	1.95 s (3 H) 2.03 s (6 H) 1.99 s (6 H)
∞	5.97 d J _{17.2} , 3.0	$5.97 ext{ d}$ $5.30 ext{ dd}$ $J_{1/2}' 3.0 ext{ } J_{2/3}' 9.2$	5.52 dd J _{3'4'} 2.4	$J_{4',5'}$ 6.6 5.32 m $J_{4',5'}$ 5.0		$J_{S'.S''} - 12.0$	3.93 dd	6.50 s	1.12 s 1.10 s		3.82 m (2 H) 1.82 m (2 H)	2.10 m 2.59 s 2.33 s	1.92 s (3 H) 2.11 s (6 H) 2.05 s (6 H)
10	5 82 d J _{1'.2'} 8.6	5 82 d 5.61 dd J _{1.2} , 8.6 J _{2.3} , 1.6	5.55 dd J _{3'4'} 9.0	$J_{4',S''}$ 7.5 5.10 m $J_{4',S'}$ 3.6 $J_{4',S'}$ 7.3		4.25 dd J _{S',S"} 15.6	3.88 dd	6.67s 1.	1.09 s	4.10 q (2 H) 1.30 t (3 H)	1.01t (3H)	2.57 s 2 30 s	1.97 s (3 H) 2.03 s (12 H) 1.84 s (3 H)

"Recorded for a solution in CDC1, at 35.5°, 8 scale (internal Me,Si); J in Hz, with assignments verified by selective proton-decoupling. Bingle primes denote numbering of atoms on the sugar chain. Double primes are reserved for the labeling of the upfield proton (usually) of a methylene group. Heterocycle.

Signal multiplicities. d, doublet; dd. doublet; m, multiplet; q, quartet; s, singlet; and t, triplet.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(D-galacto-pentitol-1-yl)indol-4-one (3). — A solution of 2-deoxy-2-ethylamino-D-glycero-L-gluco-heptose hydrochloride⁸ (10.2 g, 37.3 mmol) in water (30 mL) was treated with 5,5-dimethyl-1,3-cyclohexanedione (5.2 g, 37.3 mmol) in 2:1 acetone-water (50 mL), and sodium carbonate (2.0 g, 18.7 mmol) was added. The mixture was kept for 13 days at room temperature, and then the acetone was evaporated under diminished pressure, to give 3 as white crystals that were collected by filtration (8.0 g, 63%). Recrystallized from 96% ethanol, it gave plates, m.p. 182–184°, [α]_D +9.5°, [α]₅₇₈ +10°, [α]₅₄₆ +12°, [α]₄₃₆ +26°, [α]₃₆₅ +64° (c 1.0, pyridine); $\lambda_{\text{max}}^{\text{EtoH}}$ 250 and 280 nm (ε_{mM} 8.90 and 6.40); ν_{max} 3600–3100 (OH), 1620 (C=O), 1550, and 1505 cm⁻¹ (C=C pyrrole).

Anal. Calc. for $C_{17}H_{27}NO_6$: C, 59.81; H, 7.97; N, 4.10. Found: C, 59.73; H, 8.15; N, 4.17. Periodate consumption: 4.16 mol.

1-Ethyl-4,5,6,7-tetrahydro-2-(D-galacto-pentitol-1-yl)indol-4-one (4). — A solution of 2-deoxy-2-ethylamino-D-glycero-L-gluco-heptose hydrochloride⁸ (11.0 g, 40.2 mmol) in water (25 mL) was treated with 1,3-cyclohexanedione (4.5 g, 40.2 mmol) in water (20 mL), and sodium carbonate (2.1 g, 20.1 mmol) was added. The mixture was kept for 20 days at room temperature, filtered to remove some solid impurities, and the filtrate concentrated under diminished pressure until crystallization occurred. The crude product (3.5 g) was filtered off, and successively washed with cold 80% methanol and acetone. The mother liquor and washings yielded several crops of crystals (total yield: 5.3 g, 42%). Recrystallized twice from 96% ethanol, it gave plates, m.p. 163–165°, [α]_D +14°, [α]₅₇₈ +15°, [α]₅₄₆ +18°, [α]₄₃₆ +36°, [α]₃₆₅ +74° (c 0.8, pyridine); λ_{max}^{EtOH} 248 and 275 nm (ε_{mM} 9.50 and 6.80); ν_{max} 1615 (C=O), 1550, and 1505 cm⁻¹ (C=C pyrrole).

Anal. Calc. for $C_{15}H_{23}NO_6$: C, 57.50; H, 7.40; N, 4.47. Found: C, 57.79; H, 7.69; N, 4.47. Periodate consumption: 3.98 mol.

4,5,6,7-Tetrahydro-6,6-dimethyl-2-(D-galacto-pentitol-1-yl)-1-propylindol-4-one (5). — A solution of a mixture of 2-deoxy-2-propylamino-D-glycero-L-gluco-and -D-glycero-L-manno-heptose hydrochloride⁸ (1.0 g, 3.5 mmol) in water (5 mL) was treated with 5,5-dimethyl-1,3-cyclohexanedione (0.5 g, 3.5 mmol) in 2:1 acetone-water (13 mL), and sodium hydrogencarbonate (0.3 g, 3.5 mmol) was added. The mixture was kept for 13 days at room temperature, and then the acetone was evaporated under diminished pressure. The resulting solution was washed with ethyl acetate (2 × 15 mL), and then extracted with 10:1 ethyl acetate-ethanol (10 × 15 mL). The extracts were evaporated to give 5 (0.2 g, 15%). Recrystallized from abs. ethanol, it had m.p. 149–150°, $[\alpha]_D$ +6°, $[\alpha]_{578}$ +6°, $[\alpha]_{546}$ +8°, $[\alpha]_{436}$ +20°, $[\alpha]_{365}$ +54° (c 0.5, pyridine); λ_{max}^{EtOH} 251 and 276 nm (ε_{mM} 10.80 and 9.00); ν_{max} 1620 (C=O), 1555, and 1510 cm⁻¹ (C=C pyrrole).

Anal. Calc. for $C_{18}H_{29}NO_6$: C, 60.82; H, 8.22; N, 3.93. Found: C, 60.52; H, 8.05; N, 3.87. Periodate consumption: 3.91 mol.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(penta-O-acetyl-D-galacto-pentitol-1-yl)indol-4-one (6). — A suspension of 3 (0.75 g, 2.2 mmol) in pyridine (5 mL) and acetic anhydride (3.8 mL) was kept for 24 h at room temperature, and then

poured into ice–water (50 mL). The solid **6** (1.15 g, 95%) was filtered off, washed with water, and dried over sodium hydroxide; m.p. 130–132°, $[\alpha]_D$ +58°, $[\alpha]_{578}$ +61°, $[\alpha]_{546}$ +71°, $[\alpha]_{436}$ +133°, $[\alpha]_{365}$ +232° (c 1.0, chloroform); ν_{max} 1740 (C=O ester), 1640 (C=O ketone), 1545, and 1510 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table I.

Anal. Calc. for $C_{27}H_{37}NO_{11}$: C, 58.79; H, 6.76; N, 2.54. Found: C, 58.67; H, 6.98; N, 2.72.

1-Ethyl-4,5,6,7-tetrahydro-2-(penta-O-acetyl-D-galacto-pentitol-1-yl)indol-4-one (7). — A suspension of 4 (0.47 g, 1.5 mmol) in pyridine (2.5 mL) and acetic anhydride (3.8 mL) was kept for 17 h at room temperature, and then poured into ice-water (50 mL). The solid 7 (0.45 g, 57%) was filtered off, washed with cold water, and dried *in vacuo* over sodium hydroxide; m.p. 129–131°, $[\alpha]_D$ +58°, $[\alpha]_{578}$ +61°, $[\alpha]_{546}$ +70°, $[\alpha]_{436}$ +134°, $[\alpha]_{365}$ +238° (c 0.8, chloroform); ν_{max} 1735 (C=O ester), 1650 (C=O ketone), 1545, and 1505 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table I.

Anal. Calc. for $C_{25}H_{33}NO_{11}$: C, 57.35; H, 6.35; N, 2.68. Found: C, 57.34; H, 6.46; N, 2.65.

4,5,6,7-Tetrahydro-6,6-dimethyl-2-(penta-O-acetyl-D-galacto-pentitol-1-yl)-1-propylindol-4-one (8). — A suspension of 5 (0.1 g, 0.3 mmol) in pyridine (0.5 mL) and acetic anhydride (0.5 mL) was kept for 24 h at room temperature, and then poured into ice-water (30 mL). The solid 8 (0.15 g, 92%) was filtered off, washed with water, and dried over sodium hydroxide. Recrystallized from ethanol-water, it had m.p. 75-76°, $[\alpha]_D$ +47°, $[\alpha]_{578}$ +49°, $[\alpha]_{546}$ +55°, $[\alpha]_{436}$ +108.5°, $[\alpha]_{365}$ +189° (c 0.4, chloroform); ν_{max} 1745 (C=O ester), 1650 (C=O ketone), 1575, and 1525 cm⁻¹ (C=C pyrrole); 1 H-n.m.r. data are given in Table I.

Anal. Calc. for $C_{28}H_{39}NO_{11} \cdot H_2O$: C, 57.62; H, 7.08; N, 2.40. Found: C, 57.87; H, 6.97; N, 2.49.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(D-manno-pentitol-1-yl)indol-4-one (9). — A solution of 2-deoxy-2-ethylamino-D-glycero-D-talo-heptose hydrochloride⁷ (5.0 g, 18.3 mmol) in water (10 mL) was treated with 5,5-dimethyl-1,3-cyclohexanedione (2.6 g, 18.3 mmol) in 2:1 acetone-water (26 mL), and sodium carbonate (0.97 g, 9.2 mmol) was added. The mixture was kept for 20 days at room temperature, and then the acetone was evaporated under diminished pressure. The resulting solution was kept for 4 days at 0°. Compound 9 was filtered off, washed successively with water, acetone, and ether, and dried over sodium hydroxide (yield 2.62 g. 42%). Recrystallized from water and then from abs. ethanol, it gave needles, m.p. 156–158°, $[\alpha]_D$ –11.5°, $[\alpha]_{578}$ –12°, $[\alpha]_{546}$ –15°, $[\alpha]_{436}$ –23°, $[\alpha]_{365}$ –34° (c 0.4, pyridine); λ_{max}^{EtOH} 250 and 280 nm (ε_{mM} 8.90 and 6.30); ν_{max} 3600–3000 (OH), 1610 (C=O), 1550, and 1505 cm⁻¹ (C=C pyrrole).

Anal. Calc. for $C_{17}H_{27}NO_6$: C, 59.81; H, 7.97; N, 4.10. Found: C, 59.98; H, 8.26; N, 3.95. Periodate consumption: 3.97 mol.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-2-(penta-O-acetyl-D-manno-pentitol-1-yl)indol-4-one (10). — A solution of 9 (0.7 g, 2 mmol) in pyridine (3 mL) and

acetic anhydride (3.5 mL) was kept for 17 h at room temperature, and then poured into ice-water (50 mL). The solid **10** (1.0 g, 98%) was filtered off, washed with cold water, and dried over sodium hydroxide. Recrystallized from ether-light petroleum, it had m.p. 119–121°, $[\alpha]_D$ –12°, $[\alpha]_{578}$ –13°, $[\alpha]_{546}$ –15°, $[\alpha]_{436}$ –35°, $[\alpha]_{365}$ –75° (c 0.6, chloroform); ν_{max} 1745 (C=O ester), 1650 (C=O ketone), 1550, and 1500 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table I.

Anal. Calc. for $C_{27}H_{37}NO_{11}$: C, 58.79; H, 6.76; N, 2.54. Found: C, 59.02; H, 7.02; N, 2.76.

4,5,6,7-Tetrahydro-6,6-dimethyl-2-(D-gluco-pentitol-1-yl)-1-propylindol-4-one (11). — A solution of a mixture of 2-deoxy-2-propylamino-D-glycero-D-ido-and D-glycero-D-gulo-heptose hydrochloride³ (1.0 g, 3.5 mmol) in water (5 mL) was treated with 5,5-dimethyl-1,3-cyclohexanedione (0.5 g, 3.5 mmol) in 2:1 acetone-water (13 mL), and sodium hydrogencarbonate (0.3 g, 3.5 mmol) was added. The mixture was kept for 13 days at room temperature, and then the acetone was evaporated under diminished pressure. The resulting solution was washed with chloroform (2 × 15 mL), and then extracted with 10:1 ethyl acetate-ethanol (10 × 15 mL). These extracts were evaporated to give 11 (0.15 g, 12%). Recrystallized from acetone, it had m.p. 117–118°, $[\alpha]_{\rm b}$ +1°, $[\alpha]_{\rm 578}$ +2°, $[\alpha]_{\rm 546}$ +2°, $[\alpha]_{\rm 436}$ +12°, $[\alpha]_{\rm 365}$ +43° (c 1.0, pyridine); $\lambda_{\rm max}^{\rm EtOH}$ 252 and 280 nm ($\varepsilon_{\rm mM}$ 10.00 and 7.90); $\nu_{\rm max}$ 1630 (C=O), 1560, and 1510 cm⁻¹ (C=C pyrrole).

Anal. Calc. for $C_{18}H_{29}NO_6$: C, 60.82; H, 8.22; N, 3.93. Found: C, 60.75; H, 8.40; N, 4.12. Periodate consumption: 4.00 mol.

1-Alkyl-4,5,6,7-tetrahydro-4-oxoindol-2-carboxaldehydes (general method). — An ice-cold suspension of the 2-(alditol-1-yl)-1-alkyl-4,5,6,7-tetrahydroindol-4-one (3 mmol) in water (10 mL) was treated with a cold, aqueous solution (14 mL) of sodium periodate (12 mmol) under continuous stirring. After a few minutes, the compound crystallized, and it was recrystallized from ethanol.

TABLE II

1H-N M R DATA" (90 MHz) FOR 12, 13, AND 14

Compound	СНО	H (heterocycle)	N-Et	N-Pr	CH ₂ (heterocycle)	CH ₃ (heterocycle)
12	9.56 s ^b	7.29 s	4.37 q		2.71 s (2 H)	1.16 s (6 H)
			1.33 t		$2.40 \mathrm{s} (2 \mathrm{H})$	
			J7.0		, ,	
13	9.56 s	7.29 s	4.36 q		2.86 t (2 H)	
			1.35 t		2.52 m (2 H)	
			J7.0		2.23 m (2 H)	
14	9.53 s	7.31 s		4.25 t	2.71 s (2 H)	1.20 s (6 H)
				1.73 m	2.40 s (2 H)	. ,
				0.93 t	, ,	

[&]quot;Recorded for a solution in CDCl₃ at 35.5°; δ scale (internal Me₄Si), J in Hz. ^bSignal multiplicities: m, multiplet; q, quartet; s, singlet; t, triplet.

1-Ethyl-4,5,6,7-tetrahydro-6,6-dimethyl-4-oxoindol-2-carboxaldehyde (12). — This was prepared from 1, 3, or 9 by the method just described; yield, 72%; m.p. 113–115°; $\lambda_{\text{max}}^{\text{EtOH}}$ 230, 254, and 294 nm (ε_{mM} 19.60, 9.10, and 18.80); ν_{max} 1640 (C=O), 1515, and 1480 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table II.

Anal. Calc. for $C_{13}H_{17}NO_2$: C, 71.21; H, 7.81; N, 6.39. Found: C, 70.99; H, 7.93; N, 6.45.

1-Ethyl-4,5,6,7-tetrahydro-4-oxoindol-2-carboxaldehyde (13). — This was prepared from 4 by the general method; yield, 82%; m.p. 140–142°; $\lambda_{\rm max}^{\rm EtOH}$ 230, 254, and 294 nm ($\varepsilon_{\rm mM}$ 19.80, 8.90, and 18.60); $\nu_{\rm max}$ 1640 (C=O), 1520, and 1485 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table II.

Anal. Calc. for $C_{11}H_{13}NO_2$: C, 69.09; H, 6.85; N, 7.33. Found: C, 68.86; H, 7.05; N, 7.32.

4,5,6,7-Tetrahydro-6,6-dimethyl-4-oxo-1-propylindol-2-carboxaldehyde (14). — This was prepared from 5 or 11 by the general method; yield, 86%; m.p. 110–111°; $\lambda_{\text{max}}^{\text{EtOH}}$ 298 nm (ε_{mM} 5.30); ν_{max} 1665 and 1650 (C=O), 1555, and 1515 cm⁻¹ (C=C pyrrole); ¹H-n.m.r. data are given in Table II.

Anal. Calc. for $C_{14}H_{19}NO_2$: C, 72.07; H, 8.21; N, 6.00. Found: C, 72.32; H, 8.54; N, 5.72.

ACKNOWLEDGMENTS

This work was supported by a grant from the C.A.I.C.T. of the Ministry of Education and Science of Spain. We thank Isabel Cruz Murillo for her expert technical assistance.

REFERENCES

- 1 A. GOMEZ SANCHEZ, E. TOLEDANO, AND M. GOMEZ GUILLEN, J. Chem. Soc., Perkin Trans. 1, (1974) 1237–1243.
- 2 F. GARCIA GONZALEZ, M. GOMEZ GUILLEN, J. A. GALBIS PEREZ, AND E. ROMAN GALAN, Carbohydr. Res., 78 (1980) 17-23.
- 3 F. GARCIA GONZALEZ, M. GOMEZ GUILLEN, J. A. GALBIS PEREZ, AND E. ROMAN GALAN, Carbohydr. Res., 80 (1980) 37-43.
- 4 J. A. GALBIS PEREZ, E. ROMAN GALAN, J. L. JIMENEZ REQUEJO, AND F. POLO CORRALES, Carbohydr. Res., 102 (1982) 111–119.
- 5 E. ROMAN GALAN, J. A. GALBIS PEREZ, AND M. A. AREVALO AREVALO, Carbohydr. Res., 116 (1983) 255–262.
- 6 J. A. Galbis Perez, J. C. Palacios Albarran, J. L. Jimenez Reouejo, M. Avalos Gonzalez, and J. M. Fernandez-Bolaños, Carbohydr. Res., 129 (1984) 131–142.
- 7 J. A. Galbis Perez, J. C. Palacios Albarran, J. L. Jimenez Requejo, M. Avalos Gonzalez, and J. M. Fernandez-Bolaños, *Carbohydr. Res.*, 131 (1984) 71–82.
- 8 J. A. Galbis Perez, J. L. Jimenez Requejo, J. C. Palacios Albarran, M. Avalos Gonzalez, and J. M. Fernandez-Bolaños, unpublished results.
- 9 N. K. RICHTMYER AND C. S. HUDSON, J. Am. Chem. Soc., 64 (1942) 1612–1613.
- 10 H. S. EL KHADEM AND Z. M. EL SHAFEI, Tetrahedron Lett., (1963) 1887-1889.
- 11 H. S. El Khadem, Carbohydr. Res., 59 (1977) 11-18.
- 12 F. GARCIA GONZALEZ, J. FERNANDEZ-BOLAÑOS, AND M. A. PRADERA DE FUENTES, An. Quím., 70 (1974) 57–59.